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Chemistry is the branch of science which deals specifically with the elements on the periodic table, as well as compounds, all of which are composed of atoms, molecules, and ions. In this quiz, we aim to boost your knowledge on a chemistry-oriented level on some everyday items such as salt and aluminum foil, as well as your expertise

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on individual elements and their chemical symbol.

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Question. 17. Define atomic mass unit.

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Answer. One atomic mass unit is a mass unit equal to exactly one twelfth ($1/12$ th) the mass of one atom of carbon -12.

Question. 18. The relative atomic mass of oxygen atom is 16. Explain its meaning. Answer.

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Figure 9.2.1 Dissolution and Precipitation

(a) When a solid is added to a solvent in which it is soluble, solute particles leave the surface of the solid and become solvated by the solvent, initially forming an unsaturated solution. (b) When the

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maximum possible amount of solute has dissolved, the solution becomes saturated. If excess solute is present, the rate at which solute particles leave ...

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because of the increase in the number of
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What is the molarity of a potassium permanganate solution if 25.0 mL of it reacts with 10.0 mL of 0.0100 M Na_2SO_3 solution in acidic solution? The following equation is NOT balanced. $\text{SO}_3^{2-} + \text{MnO}_4^- \rightarrow \text{SO}_4^{2-} + \text{Mn}^{2+}$

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Unit 9: Applications of Thermodynamics
You'll be introduced to the concept of "thermodynamic favorability" for reactions, meaning how likely they are to occur given energy changes and

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